

STIC Search Report

STIC Database Tracking Number

TO: Monique Wills Location: REM 6C21 Art Unit: 1746 May 19, 2004

Case Serial Number: 09/963800

From: Kathleen Fuller Location: EIC 1700 REMSEN 4B28 Phone: 571/272-2505 Kathleen.Fuller@uspto.gov

sei	arc	M,	ote	es l
pat it			COUNTY	





STIC Search Results Feedback Form

FIC17000

Comments:

Questions about the scope or the results of the search? Contact the E/C searcher or contact:

Kathleen Fuller, EIC 1700 Team Leader 571/272-2505 REMSEN 4B28

AA	I am an examiner in Workgroup: Example: 1713 Relevant prior art found, search results used as follows:
	102 rejection
	103 rejection
	Cited as being of interest.
	Helped examiner better understand the invention.
	 Helped examiner better understand the state of the art in their technology.
	Types of relevant prior art found:
	Foreign Patent(s)
	Non-Patent Literature (journal articles, conference proceedings, new product announcements etc.)
Þ	Relevant prior art not found:
	 Results verified the tack of relevant prior art (helped determine patentability).
	Beguitte were not useful in determining patentability or understanding the invention.



SEARCH REQUEST FORM

Scientific and Technical Information Center

Requester's Full Name: UMUQUE UUD Examiner # : 75068 Date: \$ 1700
Art Unit: 1740 Phone Number 3r 272 - 1309 Serial Number 0010(62 Mars)
Mail Box and Bldg/Room Location: 1021 Results Format Preferred (circle). PAPER DISK E-MAIL
If more than one search is submitted, please prioritize searches in order of need.
Please provide a detailed statement of the search topy; and describe an specifically as possible duralport matter to be searched. Include the efenting species or structures, seywords, prospares, accessions, and registery numbers, and combine with the counterport solidy of the invention. Define any terms that may have a special meaning. Give examples or relevant instrument, such as for a such as the control of the contro
Title of Invention. An Electro Chances Cell Leaving a conhalad electrode Invention of Market Changes of Energy.
Inventors (plats provide full names): EVIC S. ROID. DUNIS G. Fauteux
Keichi Seli
Earliest Priority Filing Date. 926 260

STAFF USES SOLV Jype of Starch
Storte Land Solve Solve

-> FILE REG

FILE 'REGISTRY' ENTERED AT 17:41:22 ON 19 MAY 2004 USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT.

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Property values tagged with IC are from the ZIC/VINITI data file provided by InfoChem.

STRUCTURE FILE UPDATES: 18 MAY 2004 HIGHEST RN 683203-75-0 DICTIONARY FILE UPDATES: 18 MAY 2004 HIGHEST RN 683203-75-0

TSCA INFORMATION NOW CURRENT THROUGH JANUARY 6, 2004

Please note that search-term pricing does apply when conducting SmartSELECT searches.

Crossover limits have been increased. See HELP CROSSOVER for details.

Experimental and calculated property data are now available. For more information enter HELP PROP at a zerow promit in the file or refer to the file summary sheet on the who have the high summary sheet with the property case org/ONDEMPED 9885878.

=> FILE HCAPLUS

FILE HEARTHS ENTERED AT 17:41:27 ON 19 MAY 2004
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FILE COVERS 1907 - 19 May 2004 VOL 140 ISS 21 FILE LAST UPDATED: 18 May 2004 (20040518/ED)

This file contains CAS Registry Numbers for easy and accurate substance identification.

=> D QUE

11 SEA FILE=REGISTRY ABB=ON (108-32-7/BI OR 1121-07-9/BI OR 12190-79-3/BI OR 21324-40-3/BI OR 24937-79-9/BI OR 4166-53-4/BI OR 5/636-10-9/BI OR 7762-42-5/BI OR 78-59-1/BI OR 672-50-4/BI OR 86-48-1/BI)

L39 6 SEA FILE-REGISTRY ABB-ON L38 AND 1-4/NR
L47 611 SEA FILE-REGISTRY ABB-ON METHYL AND OXA AND TRICYCLO AND

L48 508 SEA FILE=REGISTRY ABB=ON L47 NOT PMS/CI L49 253 SEA FILE=REGISTRY ABB=ON L48 AND 3/NR

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5/19/04
                                Page 2
T.51
              1 SEA FILE=REGISTRY ABB=ON L49 AND DEC(W)8(W) ENE
1.79
            430 SEA FILE=REGISTRY ABB=ON 197.10.1/RID
1.80
           1776 SEA FILE=REGISTRY ABB=ON 197.9.1/RID
           2460 SEA FILE-REGISTRY ABB-ON 180.93.1/RID
L83
T.R.S.
            212 SEA FILE=REGISTRY ABB=ON L83 AND 2/NR
1.86
             15 SEA FILE-REGISTRY ABB-ON L85 AND DIONE
1.87
              I SEA FILE-REGISTRY ABB-ON L86 AND C7H8O4/MF
LBB
            284 SEA FILE=REGISTRY ABB=ON (L79 OR L80) AND DIONE
L89
            208 SEA FILE-REGISTRY ABB-ON
                                          L88 AND 1(W)8(W)8(W)TRIMETHYL
L90
             37 SEA FILE=REGISTRY ABB=ON
                                          L89 AND 2/NR
L91
              3 SEA FILE=REGISTRY ABB=ON
                                          L90 AND C10H14O3/MF
1.94
              1 SEA FILE=REGISTRY ABB=ON L39 AND C5H7NO2/MF
1.96
              1 SEA FILE-REGISTRY ABB-ON L39 AND C9H14O/MF
1.97
              1 SEA FILE=REGISTRY ABB=ON L39 AND C6H8O3/MF
L99
              8 SEA FILE=REGISTRY ABB=ON L94 OR L96 OR L97 OR L51 OR L87 OR
                                                                   omsounde
                L91
T.100
           3560 SEA FILE=HCAPLUS ABB=ON 1.99
             23 SEA FILE-HCAPLUS ABB=ON L100 AND BATTER?
             11 SEA FILE=HCAPLUS ABB=ON L102 AND (ELECTRODE? OR CATHODE? OR
                ANODE?1
L105
              5 SEA FILE-HCAPLUS ABB-ON L100 AND ELECTROCHEM? (5A) CELL#
L106
             12 SEA FILE=HCAPLUS ARR=ON 1.105 OR 1.103
=> D L106 ALL HITSTR 1-12
L106 ANSWER 1 OF 12 HCAPLUS COPYRIGHT 2004 ACS on STN
ъw
     2003:279778 HCAPLUS
DN
     138:306802
ED
    Entered STN: 11 Apr 2003
     Secondary nonaqueous electrolyte battery
IN
     Morishima, Hideaki; Ota, Hideo; Yamada, Shuji
PA
    Toshiba Corp., Japan
80
     Jpn. Kokai Tokkyo Koho, 15 pp.
     CODEN: JKXXAF
    Patent
LA
     Japanese
     TCM H01M010-40
     ICS H01M004-58
     52-2 (Electrochemical, Radiational, and Thermal Energy Technology)
FAN. CNT 1
     PATENT NO.
                      KIND DATE
                                           APPLICATION NO.
                                                            DATE
     JP 2003109661
                       A2
                           20030411
                                           JP 2001-298618
                                                            20010927
PRAI JP 2001-298618
                            20010927
AB
    The bat tery has a nonag. electrolyte solution containing a cyclic organic
compound,
    having an endo bridge structure, between its cathode and
    anode. The cathode may be a substituted Li Mn oxide,
    and the cyclic compound may be a heterocyclic or condensed ring compound
```

RL: DEV (Device component use); USES (Uses) (cathodes in secondary lithium batteries with

secondary lithium battery electrolyte endo bridge cyclic compd

(electrolyte solns, containing endo bridge cyclic compds, for secondary

12190-79-3, Cobalt lithium oxide (CoLiO2) 136479-44-2, Lithium magnesium

Battery electrolytes

lithium batteries)

WILLS 09/963800

ST

electrolyte solns, containing endo bridge cyclic compds.1) 96-49-1, Ethylene carbonate 623-53-0, Ethyl methyl carbonate

21324-40-3, Lithium hexafluorophosphate RL: DEV (Device component use); USES (Uses)

(electrolyte solns, containing endo bridge cyclic compds, for secondary lithium batteries)

826-62-0 24327-08-0 RL: MOA (Modifier or additive use); USES (Uses)

(electrolyte solns, containing endo bridge cyclic compds, for secondary lithium batteries)

RL: MOA (Modifier or additive use); USES (Uses)

(electrolyte solms. containing endo bridge cyclic compds. for secondary lithium batteries) 826-62-0 HCAPLUS

RN 826-62-0 HCAPLUS
CN 4,7-Methanoisobenzofuran-1,3-dione, 3a,4,7,7a-tetrahydro- (9CI) (CA INDEX NAME)



L106 ANSWER 2 OF 12 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 2003:40243 HCAPLUS DN 138:76172

ED Entered STN: 17 Jan 2003

TI Nonaqueous secondary battery IN Murai, Tetsuya; Mukai, Hiroshi PA Japan Storage Battery Co., Ltd., Japan

SO Eur. Pat. Appl., 18 pp. CODEN: EPXXDW

DT Patent LA English

IC ICM H01M010-40

CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology) FAN.CNT 1

FATENT NO. KIND DATE APPLICATION NO. DATE
FI EP 1276165 A1 20030115 EP 2002-15551 20020711
R: AT, BE, CM, DE, DK, ES, FR, GB, GR, JT, LI, LU, NL, SE, MC, PT,

IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, SK
JP 2003031259 A2 20001131 P2 2001-211767 2001017
JP 2003151623 A2 20001523 JP 2001-348541 20011114
CN 1398013 A 20030219 CN 2002-140953 20020771

US 2002-192688 20020711

US 2003054259 A1 20030320 PRAI JP 2001-211767 A 20010712 JP 2001-348541 A 20011114 OS MARPAT 138:76172

GI MARPAT I

- A8 A nonag accomdary cell includes the following elements a postelectrode capable of absorbing and releasing lithium; an engelectrode capable of absorbing and releasing lithium; and a nonagelectrode capable of absorbing and releasing lithium; and a nonagtherein wherein the electrolyte contains a vinyl ethylene carbonate of
 represented by the general formia (I); wherein Ri, R2, R3, R4, R8, and R6
 i to 4 carbon atoms, and furthernore contains at naily acoup having from
 from the group consisting of vinylene carbonate, a cyclic sulformia caid
 from the group consisting of vinylene carbonate, a cyclic sulformia caid
 from the group consisting of vinylene carbonate, a cyclic sulformia caid
 from the group consisting of vinylene carbonate, and an add analytide.
- IT Anhydrides
- RL: MOA (Modifier or additive use); USES (Uses)
- (cyclic; nonaq. electrolyte lithium secondary battery)
 IT Sulfonic acids, uses
- RL: DEV (Device component use); USES (Uses)
 (esters, cyclic; nonaq. electrolyte lithium secondary battery
- IT Secondary batteries
 (lithium; nonaq, electrolyte lithium secondary battery)
- [lithium; nonaq. electrolyte lithium secondary battery IT Battery electrolytes
 - (nonaq, electrolyte lithium secondary battery)

 Carbonaceous materials (technological products)

 RL: DEV (Device component use); USES (Uses)
- RL: DEV (Device component use), USES (Uses)
 (nonaq. electrolyte lithium secondary battery)
 IT Lactones
- RL: MOA (Modifier or additive use); USES (Uses) (nonaq. electrolyte lithium secondary battery)
- IT 96-48-0, y-Butyrolactone 96-49-1, Ethylene carbonate 105-58-8, Diethyl carbonate 342-32-9, Dibutyl carbonate 616-38-6, Dimethyl carbonate 623-35-0, Ethyl methyl carbonate 1423-07-9, Lithium tetrafluoroborate 21324-40-3, Lithium hexafluorobphosphate RL: DEV (Device component use): USES [Uses]
- Ionas, electrolyte lithium secondary hattary?
 IT 83-42-7, Cyclobexenferabowylic aniydride 85-48-8, 4-Cyclobexenferabowylic aniydride 85-48-8, 4-Cyclobexenferabowylic aniydride 8108-31-6. Nalice anhydride, uses 108-31-6. Nalice anhydride, uses 108-35-4, Glutaric anhydride 616-02-4, Citraconic anhydride 826-62-0, 5-Noborneme-7, 3-dicarbowylic anhydride 827-36-6, Vurlenc earbonate 1120-71-4,
- dicarboxylic anhydride 872-36-6, Vinylene carbonate 1120-71-4, 1.3-Propensultone 1131-15-3, Phenylauconic anhydride 1633-83-6, 1.4-butanesultone 2426-02-0, 3, 4,5,6-TITMANTOPHYMALIC ANNURIDE 2459-96-4, 2-Phenylyllutoric anhydride 3289-23-4 427-96-7, Vanyl 3289-23-4 427-96-7, Vanylauconic and Sanglauconic Albary Sulfutic acid, ester, cyclic 47878-91-7, Ethylauconic 478-96-90, Sulfutic acid, ester, cyclic 47878-91-7, Ethylauconic 478-91-91, RE: MOM, Modifieer or additive use); USES (Ugea)

(nonag. electrolyte lithium secondary battery) RE.CNT 14 THERE ARE 14 CITED REFERENCES AVAILABLE FOR THIS RECORD

(1) Anon; PATENT ABSTRACTS OF JAPAN 1992, V016(312), PE-1230 (2) Anon; PATENT ABSTRACTS OF JAPAN 2001, V2000(16)

(3) Anon; PATENT ABSTRACTS OF JAPAN 2002, V2002(06) (4) At Battery Kk; JP 2002042865 A 2002 HCAPLUS

(5) Hinohara, A; JP 2002158035 A 2002 HCAPLUS

(6) Ken-Ichi, I; WO 02056408 A 2002 HCAPLUS

(7) Matsushita Electric Ind Co Ltd; WO 0103228 A 2001 HCAPLUS

(8) Matsushita Electric Ind Co Ltd; EP 1146586 A 2001 HCAPLUS (9) Matsushita Electric Ind Co Ltd; EP 1174940 A 2002 HCAPLUS (10) Matsushita Electric Ind Co Ltd; EP 1199765 A 2002 HCAPLUS

(11) Mitsubishi Chem Corp; JP 2002190316 A 2002 HCAPLUS

(12) Mitsubishi Chemicals Corp; JP 2001006729 A 2001 HCAPLUS (13) Sanyo Electric Co Ltd; JP 04087156 A 1992 HCAPLUS

(14) Satoh, A; US 2002086216 A1 2002 826-62-0. 5-Norbornene-2,3-dicarboxylic anhydride RL: MOA (Modifier or additive use); USES (Uses) (nonaq. electrolyte lithium secondary battery)

826-62-0 HCAPLUS 4,7-Methanoisobenzofuran-1,3-dione, 3a,4,7,7a-tetrahydro- (9CI) (CA INDEX NAME)



L106 ANSWER 3 OF 12 HCAPLUS COPYRIGHT 2004 ACS on STN DM 2002:716911 HCAPLUS

ΠN 137:235278

ED Entered STN: 20 Sep 2002

Process for fabricating continuously coated electrodes on a porous current collector and battery designs incorporating the electrodes

Gan, Hong; Takeuchi, Esther S.; Rubino, Robert S. PA

U.S. Pat. Appl. Publ., 12 pp.

CODEN: USXXCO DТ

LA English IC ICM H01M004-54

ICS H01M004-58; H01M004-50; H01M004-52; H01M004-62; H01M004-70 429241000

52-2 (Electrochemical, Radiational, and Thermal Energy Technology) FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE US 2002132167 A1 US 2002-96040 20020312 JP 2002367601 A2 JP 2002-118546 20020314

PRAI US 2001-276098P P 20010315 The invention is directed to an electrochem, cell having at least one of its electrodes produced by conting a slurgy mixture of an active material, possibly a conductive additive, and a binder dispersed in a solvent and contacted to a perforated current binder dispersed in a solvent and contacted to a perforated current one of the contact of the contact of the current collector is not the contact of the current collector to block the perforations as the current collector is being collector to block the perforations as the current collector is being different active material is coated to the openie size of the collector, either as a slurry, a pressed powder, a pellet or a free standing sheet. An example of this is a cathode having a contact collector, either as slurry, a pressed powder, a pellet or a free standing sheet. An example of this is a cathode having a

T battery continuously coated electrode porous current collector

IT Fluoropolymers, uses Natural rubber, uses

Polyimides, uses RL: MOA (Modifier or

RL: MOA (Modifier or additive use); USES (Uses) (binder; process for fabricating continuously coated electrodes on porous current collector and battery designs incorporating electrodes)

IT Secondary batteries

(lithium; process for fabricating continuously coated electrodes on porous current collector and battery designs incorporating electrodes.

IT Battery electrodes

Primary batteries Secondary batteries

(process for fabricating continuously coated electrodes on porous current collector and battery designs incorporating electrodes)

IT Carbonaceous materials (technological products)
RL: DEV (Device component use); USES (Uses)

(process for fabricating continuously coated electrodes on porous current collector and battery designs incorporating electrodes)

IT Alloys, uses RL: TEM (Techni

RL: TEM (Technical or engineered material use); USES (Uses) (substrate; process for fabricating continuously coated electrodes on porous current collector and battery designs incorporating electrodes)

IT 74-85-10, Ethylene, fluorinated 115-07-10, Propylene, fluorinated 9002-88-40, Ptfe 9002-88-4, Polyethylene 9003-07-0, Polypropylene 24937-79-9, Pvdf RL: MGO (Modifier or additive use); USES (Uses)

(binder; process for fabricating continuously coated electrodes on porous current collector and battery designs incorporating electrodes)

IT 313-31-9, Manganese dioxide, uses 1344-70-3, Copper oxide 7782-42-5, Graphite, uses 11105-92-5, Silver vandium oxide 1115-78-9, Copper sulfide 11126-12-8, Iron sulfide 12031-65-1, Lithium nickel oxide 1inio2 12039-31-33, Tittanium sulfide (1752) 12057-17-9, Lithium nanganese oxide limn20d 12068-85-8, Iron disulfide 12190-79-3, Cobalt lithium oxide colio2 1789-09-2, Copper vanadium oxide 51311-17-2, Carbon fluoride 13134-56-4, Cobalt lithium nickel oxide 181163-66-4, Coc. 9211-80,0080

- RL: DEV (Device component use); USES (Uses) (process for fabricating continuously coated electrodes on porous current collector and battery designs incorporating
- IT 68-12-2 Communications of Pa-5a-1, Inophocome 78-93-3. Methyl ethyl ketone, uses 108-88-3, Toluene, uses 108-94-1, Cyclhenvanone, uses 127-19-5, Dimethylacetamide 872-50-4, n-Methyl-2-pyrrolidone, uses 7732-18-5, Water, uses

RL: TEM (Technical or engineered material use); USES (Uses) (Solvent; process for fabricating continuously coated electrodes on porous current collector and battery

designs incorporating electrodes)
7428-90-5, Aluminum, uses 7440-02-0, Nickel, uses 7440-32-6, Titanium,
uses 7440-84-4, Cobalt, uses 7440-50-8, Copper, uses 12597-68-1,
Stainless steel, uses
Ri: TEM (Technical or engineered material use); USES (Uses)

(substrate; process for fabricating continuously coated electrodes on porous current collector and battery

designs incorporating electrodes)
T 78-59-1, Isophorone

7 /8-39-1, Isoponorone RL: TEM (Technical or engineered material use); USES (Uses) (solvent; process for fabricating continuously coated electrodes on porous current collector and battery designs incorporating electrodes)

RN 78-59-1 HCAPLUS CN 2-Cyclohexen-1-one, 3,5,5-trimethyl- (8CI, 9CI) (CA INDEX NAME)

L106 ANSWER 4 OF 12 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 2002:595389 HCAPLUS DN 137:143072

D Entered STN: 09 Aug 2002 T Secondary battery having a controlled electrode

surface
IN Kolb, Eric S.; Fauteux, Denis G.; Seki, Keiichl

DSA U.S. Pat. Appl. Publ., 18 pp., Cont.-in-part of U.S. Ser. No. 362,147, abandoned.
CODEN: USYNO

DT Patent LA English

IC ICM H01M004-62 NCL 429212000

NCL 429212000 CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology) FAN.CNT 2

PATENT NO. KIND DATE APPLICATION NO. DATE
US 2002106560 Al 20020808 US 2001-963800 20010926

PRAI US 1999-362147 B2 19990728

OS MARPAT 137:143072

AB The present invention relates to an electrochem. cell

having a controlled electrode surface, comprising: a first electrode and a second electrode wherein at least one of the first and second electrodes has a carbonaceous surface; an

electrolyte containing at least one solvent an additive associated with the carhonacous surface of at least one of the first and second electrodes, wherein the additive comprises a compound having a mol. weight of 7105.

ST battery controlled electrode surface; electrolyte passivating additive battery

T Secondary batteries
(lithium; secondary battery having controlled

electrode surface)
IT Battery anodes

Battery cathodes

Battery electrolytes (secondary battery having controlled electrode

surface) IT Carbon black, uses

Fluoropolymers, uses RL: MDA (Modifier or additive use); USES (Uses)

(secondary battery having controlled electrode surface)

IT 96-49-1, Ethylene carbonate 108-32-7, Propylene carbonate 7782-42-5, Graphite, uses 12190-79-3, Cobalt lithium oxide colio2 21324-40-3, Lithium hexafiluorophosphate

RL: DEV (Device component use); USES (Uses)

(secondary battery having controlled electrode surface)

T 78-59-1, Isophorone 1121-07-9 4166-53-4, 4-Methyl-tetrahydropyran-2,6-dione 24937-79-9, Polyvinylidene fluoride 57636-10-9

RL: MOA (Modifier or additive use); USES (Uses) (secondary battery having controlled electrode

IT 872-30-4, n-Methyl-2-pyrrolidone, uses
RE: TEM (Technical or engineered material use); USES (Uses)
(secondary battery having controlled electrode

surface) T 78-59-1, Isophorone 1121-07-9 4166-53-4,

4-Methyl-tetrahydropyran-2,6-dione RL: MOA (Modifier or addative use); USES (Uses) (secondary battery having controlled electrode

surface) RN 78-59-1 HCAPLUS

CN 2-Cyclohexen-1-one, 3,5,5-trimethyl- (8CI, 9CI) (CA INDEX NAME)

RN 1121-07-9 HCAPLUS CN 2,5-Pyrrolidinedione, 1-methyl- (9CI) (CA INDEX NAME)

RN 4166-53-4 HCAPLUS CN 2H-Pyran-2,6(3H)-dione, dihydro-4-methyl- (9CI) [CA INDEX NAME]

L106 ANSWER 5 OF 12 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 2002:487887 HCAPLUS DN 137:49707

ED Entered STN: 28 Jun 2002

TI Battery cell separator and fabrication process
IN Fabrice, Coustier: Bradford Bichard

IN Fabrice, Coustier; Bradford, Richard PA Polystor Corporation, USA

SO PCT Int. Appl., 32 pp.

CODEN: PIXXD2 DT Patent

LA English

IC ICM HOIM

CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology) Section cross-reference(s): 38

FAN.	CNT	1																
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PI		2002					2002	0627		W	0 20	01-U	\$497	7.3	2001	1220		
	WO	2002	0509	29	A	3	2003	0501										
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			co,	CR,	CU,	CZ,	DE.	DK.	DM.	D2.	EC.	EE.	ES.	FT.	GB,	GD.	GE.	GH
			GM,	HR,	HU,	ID,	IL,	IN,	IS.	JP.	KE,	KG.	KP.	KR.	KZ,	LC.	LK.	LR
			LS,	LT,	LU,	LV,	MA,	MD,	MG,	MK.	MN.	MW.	MX.	MZ.	NO,	NZ.	PH.	PI.
			PT,	RO,	RU,	SD,	SE,	SG,	SI,	SK.	SL.	TJ.	TM.	TR.	TT.	т2.	HA.	HG.
			US,	UZ,	VN,	YU,	ZA,	ZW.	AM.	AZ.	BY.	KG.	K2.	MD.	RU,	T.T.	TM	
		RW:	GH,	GM,	KE,	LS,	MN,	MZ,	SD,	SL.	SZ.	TZ.	UG.	ZM.	ZW.	AT.	RF.	CH.
			CY,	DE,	DK,	ES,	FI,	FR,	GB,	GR,	IE,	IT.	LU.	MC.	NL.	PT.	SE.	TR.
			BF,	ВJ,	CF,	CG,	CI,	CM,	GA,	GN,	GQ,	GW,	ML,	MR.	NE,	SN.	TD.	TG
		2002	1107	32	A	1	2002	0815		U:	3 20	00-7	4591) "	2000	1220	,	
	AU	2002	0416	81	A	5	2002	0701		A	J 20	02-4	1681		2001	1220		
PRAI		2000					2000	1220										
	WO	2001	-US4:	9773	W		2001	1220										

AB Provided are alternative fabrication methods and compns. for an

electrochem. cell separator. The methods of the present invention are applicable to the manufacture of polymer-cased lithium-ion secondary battery cells. They are particularly, but not exclusively, represent the secondary battery cells. They are particularly, but not exclusively represent the secondary battery cased the secondary battery cased the secondary battery cased the secondary battery cased to the secondary battery cased to the secondary battery cased to the secondary

battery cell separator fabrication process

IT Secondary battery separators

(battery cell separator and fabrication process)
IT Alcohols, uses

Aromatic hydrocarbons, uses Hydrocarbons, uses

RL: TEM (Technical or engineered material use); USES (Uses) (battery cell separator and fabrication process)

IT Fluoropolymers, uses Polyoxyalkylenes, uses

ST

Polyurethanes, uses RL: MOA (Modifier or additive use); USES (Uses)

(binder; battery cell separator and fabrication process)
IT Solvents

(chlorinated; battery cell separator and fabrication process)
IT Glycols, uses
RE: TEM (Technical or engineered material use); USES (Uses)

(ethers; battery cell separator and fabrication process)
IT Ethers, uses
RL: TEM (Technical or engineered material use); USES (Uses)

[glycol; battery cell separator and fabrication process]
IT Secondary batteries
[lithium; battery cell separator and fabrication process]
IT 79-389

RL: MOA (Modifier or additive use); USES (Uses)

(battery cell separator and fabrication process) 56-23-5, Carbon tetrachloride, uses 64-17-5, Ethanol, uses 67-56-1, Methanol, uses 67-64-1, Acetone, uses 67-68-5, Dmso, uses Dmf, uses 71-43-2, Benzene, uses 78-40-0, Triethyl phosphate 78-59-1, Isophorone 78-93-3, Methyl ethyl ketone, uses 79-01-6, Trichloroethylene, uses 96-48-0, Butyrolactone 96-49-1, Ethylene carbonate 105-58-8, Diethyl carbonate 108-10-1, Methyl isobutyl ketone 108-21-4, Isopropyl acetate 108-32-7, Propylene carbonate 108-83-8, Diisobutyl ketone 108-88-3, Toluene, uses 108-90-7, Chlorobenzene, uses 108-94-1, Cyclohexanone, uses 109-66-0. Pentane, uses 109-99-9, Thf, uses 110-54-3, Hexane, uses 112-15-2, Carbitol acetate 123-42-2, Diacetone alcohol 123-86-4, n-Butyl acetate 124-18-5, Decane 127-18-4, Tetrachloroethylene, uses 127-19-5, Dimethyl acetamide 131-11-3, Dimethyl phthalate 141-97-9, Ethyl acetoacetate 512-56-1, Trimethyl phosphate 616-38-6, Dimethyl carbonate 632-22-4, Tetramethyl urea 872-50-4, n-Methylpyrrolidone, uses 1330-20-7, Xylene, uses

RL: TEM (Technical or engineered material use); USES (Uses) (battery cell separator and fabrication process)

T 116-15-4, Hexafluoropropylene 2274-11-5 9002-84-0, Ptfe 9003-05-8,

Polyacrylamide 9003-20-7, Polyvinylacetate 9003-21-8, Polymethylacrylate 9003-39-8, Polyvinylpyrrolidone 24937-79-9, Pvdf 25014-41-9, Polyacrylonitrile 25322-68-3, Peo RL: MOA (Modifier or additive use); USES (Uses)

(binder; battery cell separator and fabrication process) 78-59-1. Isophorone RL: TEM (Technical or engineered material use); USES (Uses)

(battery cell separator and fabrication process) RN 78-59-1 HCAPLUS

CN 2-Cyclohexen-1-one, 3,5,5-trimethyl- (8CI, 9CI) (CA INDEX NAME)



L106 ANSWER 6 OF 12 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 2002:466583 HCAPLUS

DN 137:35545

Entered STN: 21 Jun 2002 Use of heat-treated electrodes containing a polyamic acid-PVDF binder mixture

TM Palazzo, Marcus; Takeuchi, Esther S. DЪ

80 U.S. Pat. Appl. Publ., 14 pp. CODEN: USXXCO

Patent

LA English ICM H01M004-62

NCL 429217000

52-2 (Electrochemical, Radiational, and Thermal Energy Technology) Section cross-reference(s): 38

FAN. CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE -----US 2002076611 A1 20020620 US 2001-995202 20011127 EP 1221732 A2 20020710 EP 2001-310020 EP 1221732 A3 20020717

R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR JP 2002260668 JP 2001-402306 20011129

A2 20020913 PRAI US 2000-253972P Р 20001129

A mixture of polymeric binders that is insol. in nonag, organic electrolytes activating alkali metal or alkali metal ion electrochem. cells, is described. The mixed binder formulation provides electrodes that are flexible and non-brittle, and cells

incorporating the electrodes are dischargeable at elevated temps. A preferred binder formulation is a mixture of polyvinylidene and polvimide binders.

ST battery heat treated electrode polyamic acid PVDF binder mixt

Polyamides, uses

KATHLEEN FULLER EIC 1700 REMSEN 4B28 571/272-2505

Polyazomethines Polycarbonates, uses Polyesters, uses

Polyethers, uses Polyketones

IT

IT

TT

RL: MOA (Modifier or additive use); USES (Uses)
(binder, fluorinated; use of heat-treated electrodes containing

polyamic acid-PVDF binder mixture)

Fluoropolymers, uses

RL: MOA (Modifier or additive use); USES (Uses) (binder; use of heat-treated electrodes containing polyamic

acid-PVDF binder mixture)
Fluoropolymers, uses

RL: MOA (Modifier or additive use); USES (Uses) (flexible, rubber; use of heat-treated elect

(flexible, rubber; use of heat-treated electrodes containing polyamic acid-PVDF binder mixture)

IT Epoxy resins, uses

RL: MOA (Modifier or additive use); USES (Uses)

(fluorinated, binder; use of heat-treated electrodes containing polyamic acid-PVDF binder mixture)

IT Thermoplastic rubber

RL: MOA (Modifier or additive use); USES (Uses) (fluorinated, rubber; use of heat-treated electrodes containing polyamic acid-PVDF binder mixture.

IT Silicone rubber, uses

RL: MOA (Modifier or additive use); USES (Uses)

(fluorine-containing, binder; use of heat-treated electrodes containing polyamic acid-PVDF binder mixture) Acetals

RL: MOA (Modifier or additive use); USES (Uses)

(formals, polymers, binder, fluorinated, use of heat-treated electrodes containing polyamic acid-PVDF binder mixture) T Fluoro rubber

RL: MOA (Modifier or additive use); USES (Uses) [hexafluoropropene-tetrafluorosthylene-vinylidene fluoride, binder; use of heat-treated electrodes containing polyamic acid-PVDF binder mixture.

T Fluoro rubber RL: MOA (Modifier or additive use); USES (Uses)

(hexafluoropropene-vinylidene fluoride, binder; use of heat-treated electrodes containing polyamic acid-PVDF binder mixture)

IT Secondary batteries (11thium; use of heat-treated electrodes containing polyamic acid-PVDF binder mixture)

IT Epoxy resins, uses

RL: MOA (Modifier or additive use); USES (Uses) (methacrylates, fluorinated, binder; use of heat-treated electrodes containing polyamic acid-PVDF binder mixture) Heterocyclic compounds

RL: MOA (Modifier or additive use); USES (Uses)

(nitrogen, five-membered, polymers, binder, fluorinated; use of heat-treated electrodes containing polyamic acid-PVDF binder mixture)

IT Perfluore compounds Vinvl compounds, uses

RL: MOA (Modifier or additive use); USES (Uses)

(perfluoroalkyl vinyl ether polymers, tetrafluoroethylene copolymer with, binder; use of heat-treated electrodes containing polyamic acid-PVDF binder mixture)

Fluoro rubber

IT

IT

IT

RL: MOA (Modifier or additive use); USES (Uses)

(perfluoroalkyl vinyl ether-tetrafluoroethene, binder; use of heat-treated electrodes containing polyamic acid-PVDF binder mixture)

Fluoro rubber RL: MOA (Modifier or additive use); USES (Uses)

(perfluoroalkyl vinyl ether-tetrafluoroethylene-vinylidene fluoride, binder; use of heat-treated electrodes containing polyamic acid-PUPF binder mixture]

IT Ethers, uses

RL: MOA (Modifier or additive use); USES (Uses) (perfluorcalky) vinyl, polymers, tetrafluorcethylene copolymer with, binder; use of heat-treated electrodes containing polyamic acid-PVDF binder mixture!

T Synthetic rubber, uses

RL: MOA (Modifier or additive use); USES (Uses)
(phosphazene, fluorinated, rubber; use of heat-treated
electrodes containing polyamic acid-PVDF binder mixture)

IT Fluoro rubber RL: MOA (Modifier or additive use); USES (Uses)

(propylene-tetrafluoroethylene, binder; use of heat-treated electrodes containing polyamic acid-PVDF binder mixture)

Fluoro rubber Rl: MOA (Modifier or additive use); USES (Uses) [silicone, binder; use of heat-treated electrodes containing

polyamic acid-PVDF binder mixture| Fluoro rubber RL: MOA (Modifier or additive use); USES (Uses)

(tetrafluoroethylene-vinylidene fluoride, binder; use of heat-treated electrodes containing polyamic acid-PVDF binder mixture)

IT Battery electrodes

Binders Primary batteries

Secondary batteries
(use of heat-treated electrodes containing polyamic acid-PVDF

binder mixture)
IT Carbonaceous materials (technological products)
Polyamic acids

Polyimides, uses RL: MOA (Modifier or additive use); USES (Uses)

(use of heat-treated electrodes containing polyamic acid-PVDF binder mixture)

II 116-14-3D, Tetrafluoroethylene, copolymer with perfluoroethylaklyl vinyl ether 9002-83-9, Polybiorortifluoroethylene 9002-83-0, Peter 9011-17-0, Rexafluoroptopylene-vinylidene fluoride copolymer 24937-79-9, Peter 2498-06-64, Polybiryl fluoroethylene 24981-14-1, Polybiryl fluoroethylene 24981-14-1, Polybiryl fluoroethylene copolymer 2507-112-2, Rexafluoroptopylene-tetrafluoroethylene copolymer 2507-14-5-3, Ethylene-chioroftifluoroethylene copolymer 2507-14-5-3,

Polyhexaflucropropylene 49717-97-7D, 2-Propenoic acid, 2-methylion(1-), homopolymer, fluorinated 64239-72-1, 2-Propenoic acid, 2-fluoro-homopolymer 149643-29-8, Fluoroethylene-vinyl ether copolymer 437609-78-40, florinate

RL: MOA (Modifier or additive use); USES (Uses) (binder; use of heat-treated electrodes containing polyamic acid-PVPF binder mixture.

IT 108-32-7, Propylene carbonate 110-71-4 11105-02-5, Silver vanadium oxide 29935-35-1. Lithium hexafluoroarsenate RL: DEV (Device component use); USES (Uses)

(use of heat-treated electrodes containing polyamic acid-PVDF hinder mixture)

IТ 68-12-2, Dmf, uses 78-59-1, Isophoron 108-88-3, Toluene, uses 108-94-1, Cyclohexanone, uses 127-19-5, n,n-Dimethylacetamide 872-50-4, n-Methylpyrrolidone, uses RL: MOA (Modifier or additive use); USES (Uses)

(use of heat-treated electrodes containing polyamic acid-PVDF binder mixture)

78-93-3, Methyl ethyl ketone, uses 7732-18-5, Water, uses RL: TEM (Technical or engineered material use); USES (Uses) (use of heat-treated electrodes containing polyamic acid-PVDF

binder mixture) 78-59-1, Isophoron

RL: MOA (Modifier or additive use); USES (Uses) (use of heat-treated electrodes containing polyamic acid-PVDF binder mixture) DM 78-59-1 HCAPLUS

CN 2-Cyclohexen-1-one, 3,5,5-trimethyl- (8CI, 9CI) (CA INDEX NAME)

L106 ANSWER 7 OF 12 HCAPLUS COPYRIGHT 2004 ACS on STN 2002:407256 HCAPLUS

DN 137:8606 FD Entered STN: 31 May 2002

TT Nonaqueous electrolyte solution and secondary battery using the

TN Hinohara, Akio; Matsuoka, Osamu PA Mitsui Chemicals Inc., Japan Jpn. Kokai Tokkyo Koho, 15 pp.

DT Patent LA Japanese

CODEN: JKXXAF ICM H01M010-40

ICS H01M004-02; H01M004-58 CC

52-2 (Electrochemical, Radiational, and Thermal Energy Technology) FAN. CNT 1 PATENT NO. KIND DATE

APPLICATION NO. DATE JP 2002158035 A2 20020531 JP 2000-353543 20001120 PRAI JF 2000-353543

The electrolyte solution has reduction peak intensities ≤200 μA/cm2 between 0.3-0.6 V at 25°, on its 1st scan on its cyclovoltammogram scanned at 10 mV/s between 0 and 3 V, using a highly oriented pyrolytic graphite working electrode and a Li reference electrode. Preferably, the electrolyte solution contains additives selected compound having a norbornene structure and/or benzenesulfonic acid derivs. The battery is a secondary Li battery.

secondary lithium battery electrolyte soln cyclovoltammogram redn peak; norbornene additive lithium battery electrolyte;

benzenesulfonic acid deriv lithium battery electrolyte additive Battery electrolytes

(norbornene and benzenesulfonic acid derivative additives in nonac. electrolyte solns. for secondary lithium batteries)
96-49-1, Ethylene carbonate 623-53-0, Ethyl methyl carbonate

4427-96-7, Vinylethylene carbonate 21324-40-3, Lithium

hexafluorophosphate RL: DEV (Device component use); USES (Uses)

(norbornene and benzenesulfonic acid derivative additives in nonag. electrolyte solms. for secondary lithium batteries) 81-08-3 121-53-9D, m-Sulfobenzoic acid, dipotassium salt

IT 826-62-0 58601-47-1 RL: MOA (Modifier or additive use); USES (Uses)

(norbornene and benzenesulfonic acid derivative additives in nonaq. electrolyte solns. for secondary lithium batteries) TT

826-62-0 RL: MOA (Modifier or additive use); USES (Uses)

(norbornene and benzenesulfonic acid derivative additives in nonaq. electrolyte solns. for secondary lithium batteries) RN 826-62-0 HCAPLUS

CN 4.7-Methanoisobenzofuran-1,3-dione, 3a,4,7,7a-tetrahydro- (9CI) (CA INDEX NAME)

ST

L106 ANSWER 8 OF 12 HCAPLUS COPYRIGHT 2004 ACS on STN an

2001:469502 HCAPLUS DN 135 - 79415

Entered STN: 29 Jun 2001 Secondary lithium batteries suppressing gas generation Yamamoto, Masaki; Seki, Kelichi; Onuki, Masamichi

DA Mitsubishi Chemical Corp., Japan 80 Jpn. Kokai Tokkyo Koho, 7 pp.

CODEN: JKXXAF DT LA Japanese

ICM H01M010-40

52-2 (Electrochemical, Radiational, and Thermal Energy Technology) FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE JP 2001176550 20010629 19991221 JP 1999-362391 19991221 PRAI JP 1999-362391

The batteries comprise nonfluidizable electrolytes at least at a place in contact with ≥1 electrodes, and the electrolytes contain ketones. The ketones inhibit generation of gases at the electrode/electrolyte interface.

lithium battery ketone electrolyte inhibition gas generation; polymer electrolyte lithium battery ketone additive

Polymer electrolytes

(battery; secondary Li batteries containing ketones in nonfluidizable electrolytes)

Secondary batteries (lithium; secondary Li batteries containing ketones in

nonfluidizable electrolytes) Battery electrolytes

(nonfluidizable; secondary Li batteries containing ketones in nonfluidizable electrolytes)

Ketones, uses

RL: MOA (Modifier or additive use): USES (Uses) (secondary Li batteries containing ketones in nonfluidizable

electrolytes) 21324-40-3P, Lithium hexafluorophosphate

RL: DEV (Device component use); PMU (Preparation, unclassified); PREP (Preparation); USES (Uses)

(electrolyte solute; secondary Li batteries containing ketones in nonfluidizable electrolytes) 96-49-1P, Ethylene carbonate 108-32-7P, Propylene carbonate

RL: DEV (Device component use); PNU (Preparation, unclassified); PREP (Preparation); USES (Uses) (electrolyte solvent; secondary Li batteries containing ketones

in nonfluidizable electrolytes) TT 7439-93-2DP, Lithium, complex with polymer, uses 173390-60-8DP, lithium complex RL: DEV (Device component use); PNU (Preparation, unclassified); PREP

(Preparation); USES (Uses) (electrolyte; secondary Li batteries containing ketones in nonfluidizable electrolytes)

78-59-1, Isophorone 93-08-3, 2'-Acetonaphthone 98-53-3 108-10-1, 4-Methyl-2-pentanone 108-94-1, Cyclohexanone, uses 110-13-4, Acetonylacetone 120-92-3, Cyclopentanone 123-19-3, 4-Heptanone 502-42-1, Cycloheptanone 539-88-8, Ethyl levulinate 583-60-8, 2-Methylcyclohexanone 589-92-4, 4-Methylcyclohexanone 3-Methylcyclohexanone 637-88-7, 1,4-Cyclohexanedione 930-30-3. 2-Cyclopenten-1-one 930-68-7, 2-Cyclohexen-1-one 2758-18-1, 3-Methyl-2-cyclopentene-1-one 4894-75-1, 4-Phenylcyclohexanone 6705-49-3, 7-0xabicyclo[4.1.0]heptan-2-one 7429-44-9, 2-Methoxycyclohexanone 79419-30-0

RL: MOA (Modifier or additive use); USES (Uses) (secondary Li batteries containing ketones in nonfluidizable

electrolytes) ΙT 78-59-1, Isophorone RL: MOA (Modifier or additive use); USES (Uses)

(secondary Li batteries containing ketones in nonfluidizable electrolytes) RN 78-59-1 HCAPLUS

CN 2-Cyclohexen-1-one, 3,5,5-trimethyl- (8CI, 9CI) (CA INDEX NAME)

L106 ANSWER 9 OF 12 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 2001:145039 HCAPLUS DN 134:195748

ED Entered STN: 28 Feb 2001

TI Batteries with carbonaceous electrodes with controlled surface, additives for surface control, and electrolytes containing the

additives
IN Kolb, Eric S.; Fauteux, Denis G.
PA Mitsubishi Chemical Corp., Japan

SO Jpn. Kokai Tokkyo Koho, 15 pp. CODEN: JKXXAP

DT Patent LA Japanese

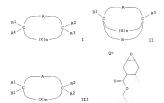
IC ICM H01M004-62

ICS H01M004-04; H01M010-40

CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)

Section cross-reference(s): 24, 27 FAN.CNT 2

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 2001057214	A2	20010227	JP 2000-225250	20000726
PRAI	US 1999-362147	A	19990728		



- AB The battery comprises 2 electrodes, at least one of which having a carbonaccous material surface treated with additives, and an electrolyte solution The additives are cyclic compds. [, II, or III [n = 0, 1, 2, 3; R1-4, Ra-c, R, R11-18 = N, C1-12 linear or branched slay!, A = C02CO, CONNACO, CO2, NNC, CHECOZ, SOZNBOCO, etc.; X = groups same as A, C1-12 linear or branched slay!, NR, O, D = groups same as A,
 - Ci-12 linear or branched alkyl, MR, O; B = groups same as A; CRIRISCRIGATION AND CRIRISCRIST, CRIRISCRICALIZATS, Q, etc.; B = H, CI-12 linear or branched alkyl, O; Electrolytes containing solvents, Li salts, and I, II, or III are also claimed. The batteries show excellent cycle characteristics and show not use generation during
- ST battery Carbonaceous electrode isophorone additive IT Battery electrolytes
 - Secondary batteries

IT

- ((bi)cyclo compds. as additives for carbonaceous electrodes
- in secondary batteries with excellent cycle characteristics)
 Battery electrodes
- (carbonaceous; (bi) cyclo compds. as additives for carbonaceous electrodes in secondary batteries with excellent cycle characteristics.
- IT 76-32-4 78-59-1, Isophorone 108-30-5, Succinic anhydride, uses 1121-07-9 2386-87-0 3425-89-6
- 4166-53-4 98546-44-2 RL: DEV (Device component use); MOA (Modifier or additive use); USES (Uses)
 - ((b1)cyclo compds. as additives for carbonaceous electrodes
 - in secondary batteries with excellent cycle characteristics) 76-32-4 78-59-1, Isophorone 1121-07-9 4166-53-4 98546-44-2
- RL: DEV (Device component use); MOA (Modifier or additive use); USES (Uses)
- ((bi)cyclo compds. as additives for carbonaceous electrodes in secondary batteries with excellent cycle characteristics)
- RN 76-32-4 HCAPLUS CN 3-Oxabicyclo(3.2.1)octane-2,4-dione, 1,8,8-trimethyl- (9CI) (CA INDEX NAME)

- RN 78-59-1 HCAPLUS
- CN 2-Cyclohexen-1-one, 3,5,5-trimethyl- (8CI, 9CI) (CA INDEX NAME)

- 1121-07-9 HCAPLUS 2,5-Pyrrolidinedione, 1-methyl- (9CI) (CA INDEX NAME)
- RN 4166-53-4 HCAPLUS CN 2H-Pyran-2,6(3H)-dione, dihydro-4-methyl- (9CI) (CA INDEX NAME)

- RN 98546-44-2 HCAPLUS
- Furo[2,3-b]furan-2,5(3H,4H)-dione, dihydro-6a-methyl- (9CI) (CA INDEX CN NAME)

L106 ANSWER 10 OF 12 HCAPLUS COPYRIGHT 2004 ACS on STN

AN 2000:814764 HCAPLUS DN

- 133:352697 ED Entered STN: 21 Nov 2000
- Method of fabricating a laminated lithium-ion secondary battery cel1
- TN Coustier, Fabrice PA
- Polystor Corporation, USA so PCT Int. Appl., 31 pp. CODEN: PIXXD2
- Patent
- LA English
- H01M006~00; H01M004-58

52-2 (Electrochemical, Radiational, and Thermal Energy Technology) Section cross-reference(s): 38 FAN CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE WO 2000069010 000069010 A1 2<u>0001116</u> WO 2000-US12445 20000505 W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CR, CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SI, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZG, SI, SK, SI, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZG, MD, RJ, TJ, TM
RM: GH, GM, KE, LS, MM, SD, SI, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG

PRAI US 1999-133057P P 19990506 Provided are alternative fabrication methods for a lithium-ion secondary n n battery cell. Briefly, the present invention provides processes

wherein a binder is applied to a battery separator after winding or stacking the electrodes. This is accomplished by soaking previously wound or stacked electrochem, structures, including pos, and neg. electrodes and a porous separator, in a solution containing a

binder. battery cell laminated fabrication; lithium battery

cell laminated fabrication Fluoropolymers, uses

Polyoxyalkylenes, uses

Polyurethanes, uses RL: TEM (Technical or engineered material use); USES (Uses) (binder; method of fabricating laminated lithium-ion secondary

battery cell) IT

Glycols, uses RL: TEM (Technical or engineered material use); USES (Uses)

(esters, solvent; method of fabricating laminated lithium-ion secondary battery cell)

Glycols, uses Glycols, uses RL: TEM (Technical or engineered material use); USES (Uses)

(ethers, solvent; method of fabricating laminated lithium-ion secondary battery cell) Ethers, uses

Ethers, uses RL: TEM (Technical or engineered material use); USES (Uses)

(qlycol, solvent; method of fabricating laminated lithium-ion secondary battery cell) Secondary batteries

(lithium; method of fabricating laminated lithium-ion secondary battery cell)

IΤ 2274-11-5, Ethylene Glycol diacrylate 9002-84-0, Ptfe 9003-05-8, Polyacrylamide 9003-20-7, Polyvinyl acetate 9003-21-8, Polymethyl acrylate 9003-39-8, Polyvinylpyrrolidone 24937-79-9, Pvdf 25014-41-9, Polyacrylonitrile 25322-68-3, Peo

RL: TEM (Technical or engineered material use); USES (Uses) (binder; method of fabricating laminated lithium-ion secondary

battery cell) 7429-90-5, Aluminum, uses

RL: DEV (Device component use); USES (Uses) (polymer laminated; method of fabricating laminated lithium-ion secondary battery cell)

56-23-5, Carbon tetrachloride, uses 64-17-5, Ethanol, uses 67-56-1. Methanol, uses 67-64-1, Acetone, uses 67-68-5, Dmso, uses 68-12-2. Dmf, uses 71-43-2, Benzene, uses 78-40-0, Triethyl phosphate 78-59-1, Isophorone 78-93-3, Methyl ethyl ketone, uses 79-01-6, Trichloroethylene, uses 96-48-0, Butyrolactone Ethylene carbonate 105-58-8, Diethyl carbonate 108-10-1, Methyl isobutyl ketone 108-21-4, Isopropyl acetate 108-32-7, Propylene carbonate 108-83-8, Diisobutyl ketone 108-88-3, Toluene, uses 108-90-7, Chlorobenzene, uses 108-94-1, Cyclohexanone, uses 109-66-0, Pentane, uses 109-99-9, Thf, uses 110-54-3, Hexane, uses 112-15-2. Carbitol acetate 123-42-2, Diacetone alcohol 123-86-4, n-Butyl acetate 124-18-5, Decane 127-18-4, Tetrachloroethylene, uses 127-19-5, Dimethyl acetamide 131-11-3, Dimethyl phthalate 141-97-9, Ethyl acetoacetate 512-56-1, Trimethyl phosphate 616-38-6, Dimethyl carbonate 632-22-4, Tetramethyl urea 872-50-4, n-Methylpyrrolidone, uses 1330-20-7, Xylene, uses

RL: TEM (Technical or engineered material use); USES (Uses) (solvent; method of fabricating laminated lithium-ion secondary

battery cell)
RE.CNT 3 THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD

RE.CNT 3 THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD

- (1) Gozdz; US 5456000 A 1995 HCAPLUS
- (2) Schmutz; US 5470357 A 1995 HCAPLUS (3) Venugopal; US 5853916 A 1998 HCAPLUS
- T 78-59-1, Isophorone

RL: TEM (Technical or engineered material use); USES (Uses)
(solvent; method of fabricating laminated lithium-ion secondary
battery cell)
RN 78-59-1 HCAPUIS

CN 2-Cyclohexen-1-one, 3,5,5-trimethyl- (8CI, 9CI) (CA INDEX NAME)

L106 ANSWER 11 OF 12 HCAPLUS COPYRIGHT 2004 ACS on STN AN 2000:219069 HCAPLUS

DN 132:239423

ED Entered STN: 05 Apr 2000

TI Secondary battery having a controlled electrode surface and associated fabrication and chemical process

IN Kolb, Eric S.; Van Buren, Martin; Fauteux, Denis G.

PA Mitsubishi Chemical Corporation, Japan SO U.S., 8 pp., Cont.-in-part of U.S. 5,853,917.

CODEN: USXXAM DT Patent

LA English

IC ICM H01M010-34 NCL 429059000

CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)

Bay you have a said

Section cross-reference(s): 72 FAN. CNT 7

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	(US 604593)7	A	20000404	US 1998-178846	19981026
	US_8053917	A	19981229	US 1997-812021	19970306
	JP 10302836	A2	19981113	JP 1998-50379	19980303
	US 6168878	B1	20010102	US 1998-208895	19981210
PRAI	US 1997-812021	A2	19970306		
	IIS T998-T78846	2.2	19981026		

An electrochem. cell includes a controlled AB

electrode surface comprising an electrode with a

carbonaceous surface, an electrolyte and a reduced additive. The reduced additive is formulated from an additive which is either soluble or insol. in the solvated electrolyte prior to reduction The invention further includes a passivating layer at the carbonaceous electrode/electrolyte interface. The passivating layer includes the additive and/or the reduced additive. This passivating layer substantially precludes contact between electrolyte solvent and the carbonaceous surface of the electrode to, in turn, substantially prevent gas formation within the cell, which would otherwise result from decomposition of the solvent upon contact with the carbonaceous surface. Also, the additive and/or the reduced additive will likewise be substantially precluded from generating a gas upon its decomposition

ST lithium battery controlled electrode surface

TT Secondary batteries

(lithium; secondary battery having controlled electrode surface and associated fabrication and chemical process)

Battery anodes (secondary battery having controlled electrode

surface and associated fabrication and chemical process) IT 7440-50-8, Copper, uses

RL: DEV (Device component use); USES (Uses) (anode grid; secondary battery having controlled

electrode surface and associated fabrication and chemical process; 7782-42-5, Graphite, uses RL: DEV (Device component use); USES (Uses)

(anode; secondary battery having controlled electrode surface and associated fabrication and chemical process! 108-32-7, Propylene carbonate 7439-93-2, Lithium, uses 29935-35-1. Lithium hexafluoroarsenate

RL: DEV (Device component use); USES (Uses)

(secondary battery having controlled electrode surface and associated fabrication and chemical process; 76-32-4 85-43-8 I00-42-5, Styrene, uses 108-30-5, uses TT 2386-87-0 4281-21-4 6053-68-5 19780-11-1 23911-25-3

RL: MOA (Modifier or additive use); USES (Uses) (secondary battery having controlled electrode surface and associated fabrication and chemical process)

RE. CNT 4 THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD

(1) Anon; JP 72-20756 1995 (2) Apon: JP 82-73700 1996

(3) Shu: US 5529859 1996 HCAPLUS

(4) Simon; US 5626981 1997 HCAPLUS TT 76-32-4

RL: MOA (Modifier or additive use); USES (Uses) (secondary battery having controlled electrode

surface and associated fabrication and chemical process)

RN 76-32-4 HCAPLUS CN 3-Oxabicyclo[3.2.1]octane-2,4-dione, 1,8,8-trimethyl- (9CI) (CA INDEX NAME)



L106 ANSWER 12 OF 12 HCAPLUS COPYRIGHT 2004 ACS on STN AN 1999:640250 HCAPLUS

DN 131:245597

ED Entered STN: 08 Oct 1999

TI Additives containing batteries and their manufacture IN Kolb, Eric S.

IN <u>Kolb</u>, Eric S. PA <u>Mitsubishi Chemical Industries Ltd.</u>, USA

SO Jpn. Kokai Tokkyo Koho, 34 pp. CODEN: JKXXAF

DT Patent

LA Japanese IC H01M010-04; H01M010-40

CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)
FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE
PI JP 11273707 A2 19991009 JP 1999-34452 19990212

PRAI US 1998-23690 19980213 AB The batteries have a 1st and a 2nd electrodes with

21 of the alentredes having a carbonaceous surface, an electrolyte containing 11 solvent associated with the carbonaceous electrode surface, and an additive in the electrolyte or either or both electrodes; where the additive includes means preventing all solvent of the control of the con

electrode surface to increase battery capacity. The batteries are prepared by adding the additive to the electrolyte in the battery and charging the battery, to react the

additive with the carbonaceous electrode surface.

ST battery additive electrolyte electrode reaction inhibitor.

IT Carbonaceous materials (technological products)
RL: DEV (Device component use); USES (Uses)

(additives for preventing electrolyte solvent-electrode reactions in lithium batteries with carbonaceous electrodes)

IT Secondary batteries

(lithium; additives for preventing electrolyte solventelectrode reactions in lithium batteries with carbonaceus electrodes)

In 108-32-7, Propylene carbonate 29935-35-1, Lithium hexafluoroarsenate Rl: DEV (Device component use); USES (Uses) (additives for preventing electrolyte solvent-electrode)

(additives for preventing electrolyte solvent-electrode reactions in lithium batteries with carbonaceous

electrodes)

T 98546-44-2 RL: MDA (Modifier or additive use): USES (Uses)

RI: MDA (Modifier or additive use); USES (Uses) (additives for preventing electrolyte solvent-electrode reactions in lithium batteries with carbonaceous electrodes)

IT 7782-42-5, Graphite, uses

RL: DEV [Device component use]; USES (Uses) [additives for preventing electrolyte solvent-electrode reactions in lithium batteries with graphite anodes

IT 98546-44-2

RL: MOA [Modifier or additive use], USES (Uses)
(additives for preventing electrolyte solvent-electrode reactions in lithium batteries with carbonaceous electrodes)

RN 98546-44-2 HCAPLUS
CN Furo [2,3-b] furan-2,5(3H,4H)-dione, dihydro-6a-methyl- (9CI) (CA INDEX

=> FILE REG

FILE REGISTRY ENTERED AT 17:44:07 ON 19 MAY 2004
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Property values tagged with IC are from the ZIC/VINITI data file provided by InfoChem.

STRUCTURE FILE UPDATES: 18 MAY 2004 HIGHEST RN 683203-75-0 DICTIONARY FILE UPDATES: 18 MAY 2004 HIGHEST RN 683203-75-0

TSCA INFORMATION NOW CURRENT THROUGH JANUARY 6, 2004

Please note that search-term pricing does apply when

conducting SmartSELECT searches.

Crossover limits have been increased. See HELP CROSSOVER for details,

Experimental and calculated property data are now available. For more information and or HELP PROP at an arrow promit in the file or refer to the file summary sheet to the web at:

http://www.cas.org/ONIUNE/DBSS/Tecistryss.html

=> D L99 1-8

The compounds

L99 ANSWER 1 OF 8 REGISTRY COPYRIGHT 2004 ACS on STN

N 98546-44-2 REGISTRY

CN Furo[2,3-b]furan-2,5(3H,4H)-dione, dihydro-6a-methyl- (9CI) (CA INDEX NAME)
OTHER CA INDEX NAMES:

CN Glutaric acid, 3-(1,1-dihydroxyethyl)-, di-y-lactone (6CI)

OTHER NAMES: CN NSC 151768

FS 3D CONCORD MF C7 H8 O4

CI COM

C STN Files: BEILSTEIN*, CA, CAOLD, CAPLUS, CASREACT

(*File contains numerically searchable property data)

**PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT **

9 REFERENCES IN FILE CA (1907 TO DATE)
9 REFERENCES IN FILE CAPLUS (1907 TO DATE)

1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L99 ANSWER 2 OF 8 REGISTRY COPYRIGHT 2004 ACS on STN RN 4166-53-4 REGISTRY

CN 2H-Pyran-2,6(3H)-dione, dihydro-4-methyl- (9CI) (CA INDEX NAME)

OTHER CA INDEX NAMES: CN Glutaric anhydride, 3-methyl- (6CI, 7CI, 8CI) OTHER NAMES:

CN B-Methylglutaric anhydride

CN B-Methylglutaryl anhydride CN 3-Methylglutaric anhydride

CN 4-Methyltetrahydropyran-2,6-dione

FS 3D CONCORD DR 114912-73-1

DR 114912-73-1 MF C6 H8 O3

MF C6 H8 O3 CI COM

LC STN Files: BELISTEIN-, BIOBUSINESS, CA, CAOLD, CAPIUS, CASREACT, CHEMCATS, CHEMINFORMEX, CHEMIST, CSCHEM, MODOC: IFICOB, IFIPAT, IFIUDB, SPECINFO, TOXCENTER, USPATZ, USPATFUL, (*File contains numerically searchable property data)

Other Sources: EINECS**

(**Enter CHEMLIST File for up-to-date regulatory information)

PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT

147 REFERENCES IN FILE CA (1907 TO DATE)

1 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA 147 REFERENCES IN FILE CAPLUS (1907 TO DATE) 9 REFERENCES IN FILE CADLD (PRIOR TO 1967)

L99 ANSWER 3 OF 8 REGISTRY COPYRIGHT 2004 ACS on STN RN 1121-07-9 REGISTRY

RN 1121-07-9 REGISTRY CN 2,5-Pyrrolidinedione, 1-methyl- (9CI) (CA INDEX NAME)

OTHER CA INDEX NAMES: CN Succinimide, N-methyl- (6CI, 7CI, 8CI)

OTHER NAMES: CN 1-Methyl-2,5-pyrrolidinedione

CN N-Methyl-2,5-pyrrolidinedione CN N-Methyl-2,5-pyrrolidinedione

CN N-Methylsuccinimide

FS 3D CONCORD

MF C5 H7 N O2 CI COM

STN Files: BELISTEIN*, BIOSIS, CA, CAOLD, CAPLUS, CASREACT, CHEMMATS, CHEMINORMEX, CHEMINIST, CSCHEM, DETHEMM*, GMELIN*, NODOC*, FFICDB, IFIPAT, IFIUDB, FFECS*, SPECINFO, TOXCENTER, USPATZ, USPATFULL (*File contains numerically searchable property data).

Other Sources: DSL**

(**Enter CHEMLIST File for up-to-date regulatory information)

PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT

242 REFERENCES IN FILE CA (1907 TO DATE)

1 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA

244 REFERENCES IN FILE CAPLUS (1907 TO DATE) 26 REFERENCES IN FILE CAOLD (PRIOR TO 1967)

L99 ANSWER 4 OF 8 REGISTRY COPYRIGHT 2004 ACS on STN RN 826-62-0 REGISTRY

CN 4,7-Methanoisobenzofuran-1,3-dione, 3a,4,7,7a-tetrahydro- (9CI) (CA INDEX NAME) OTHER CA INDEX NAMES:

CN 5-Norbornene-2,3-dicarboxylic anhydride (6CI, 8CI)

OTHER NAMES: CN 2-Norbornene-5,6-dicarboxylic anhydride

CN 3,6-Endomethylenephthalic anhydride, 1,2,3,6-tetrahydro-CN 3,6-Endomethylenetetrahydrophthalic anhydride CN 3.6-Methano-4-cycloheene-1,2-dicarboxyllic acid anhydride

CN 3,6-Methylene-1,2,3,6-tetrahydrophthalic anhydride

CN 4-0xatricyclo[5.2.1.02,6]dec-8-ene-3,5-dione CN 5-Norbornene-2,3-dicarboxylic acid anhydride

CN b-Morpornene-2,3-dicarpoxylic and annydride
CN bicyclo[2.2.1]hept-5-ene-2,3-dicarpoxylic anhydride
CN cis-3,6-Endomethylene-1,2,3,6-tetrahydrophthalic anhydride

CN cis-3,6-Endomethylene-1,2,3,6-tetranydrop CN Endomethylenetetrahydrophthalic anhydride

CN HIMIC CN NSC 3999

FS 3D CONCORD DR 66075-60-3

MF C9 H8 O3 CI COM LC STN Files:

STM Files: BELISTEIN*, CA, CAOLD, CAPUUS, CASREACT, CHEMCATS, CHEMINFORDRA, CHEMIST, CSCHEM, GMELIN*, BODOC*, FIFCDS, FIFFAT, IFIUDB, MSDS-OHS, RTECS*, SPECINO, TOXCENTER, USPATZ, USPATFULL (FYIG contains numerically searchable property data)

Other Sources: DSL**, EINECS**, TSCA**

(**Enter CHEMLIST File for up-to-date regulatory information)

PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT

532 REFERENCES IN FILE CA (1907 TO DATE)
130 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA

534 REFERENCES IN FILE CAPLUS (1907 TO DATE) 25 REFERENCES IN FILE CAOLD (PRIOR TO 1967)

L99 ANSWER 5 OF 8 REGISTRY COPYRIGHT 2004 ACS on STN RN 595-31-3 REGISTRY

CN 3-Oxabicyclo[3.2.1]octane-2,4-dione, 1,8,8-trimethyl-, (18,5R)-(9CT) (CA INDEX NAME) OTHER CA INDEX NAMES:

OTHER CA INDEX NAMES: CN 1,3-Cyclopentanedicarboxylic anhydride, 1,2,2-trimethyl-, (+)- (8CI) CN 3-Oxabicyclef3.2.1]octane-2.4-dione, 1,8,8-trimethyl-, (18)-

OTHER NAMES: CN (-)-Camphoric anhydride CN L-Camphoric anhydride

FS STEREOSEARCH MF C10 H14 O3

MF C10 H14 03 LC STN Files: BELLSTEIN*, CA, CAPLUS, CASREACT, CHEMLIST, HODOC*, IFICDB, IFIPAT, IFIUDB, RTECS*, USPATZ, USPATFULL

(*File contains numerically searchable property data)
Other Sources: EINEC**
(**Enter CHEMLIST File for up-to-date regulatory information)

Absolute stereochemistry.

**PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT **

11 REFERENCES IN FILE CA (1907 TO DATE) 11 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L99 ANSWER 6 OF 8 REGISTRY COPYRIGHT 2004 ACS on STN

L99 ANSWER 6 OF 8 REGISTRY COPYRIGHT 200-RN 595-29-9 REGISTRY

CN 3-Oxabicyclo[3.2.1]octane-2,4-dione, 1,8,8-trimethyl-, (1R,58)-(9CI) (CA INDEX NAME)

(9CI) (CA INDEX NAME)
OTHER CA INDEX NAMES:

CN 1,3-Cyclopentanedicarboxylic anhydride, 1,2,2-trimethyl-, (-)- (8CI)
CN 3-Oxabioyclo(3.2.1)octane-2,4-dione, 1,8,8-trimethyl-, (1R)OTHER NAMES:

CN D-(+)-Camphoric anhydride CN D-Camphoric anhydride

CN D-Campnoric annyd

FS STEREOSEARCH

MF C10 H14 O3 CI COM

LC STN Files: BELLSTEIN*, CA, CAPLUS, CASREACT, CHEMCATS, CHEMLIST, CSCHEM, IFICDB, IFIPAT, FIUDB, USPATZ, USPATFULL (*File contains numerically searchable property data)

Other Sources: BINECS**
(**Enter CHEMLIST File for up-to-date regulatory information)

Absolute stereochemistry.

**PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT **

KATHLEEN FULLER EIC 1700 REMSEN 4B28 571/272-2505

27 REFERENCES IN FILE CA (1907 TO DATE) 27 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L99 ANSWER 7 OF 8 REGISTRY COPYRIGHT 2004 ACS on STN RN 78-59-1 REGISTRY

CN 2-Cyclohexen-1-one, 3,5,5-trimethyl- (8CI, 9CI) (CA INDEX NAME)

OTHER NAMES:

CN a-Isophoron

CN a-Isophorone CN 1,1,3-Trimethyl-3-cyclohexene-5-one

CN 1,5,5-Trimethyl-3-oxocyclohexene

CN 1-Cyclohexen-3-one, 1,5,5-trimethyl-CN 3,5,5-Trimethyl-2-cyclohexen-1-one

CN 3,5,5-Trimethyl-2-cyclohexene-1-one CN 3,5,5-Trimethyl-2-cyclohexenone

CN Isoacetophorone

CN Isoforon CN Isophoron

CN Isophorone CN NSC 403657

CN NSC 403657 CN NSC 4881 FS 3D CONCORD

MF C9 H14 O

STN Files: AGRICOLA, ANASSTR, AQUIRE, BELISTERY, BIONUSINESS, BIOSIS, BIOTECHNO, OA, CANCERLIT, CADAD. CAPIUS, ACREACT, CERB, CDH. CHEMCAT, CHEMINFONKER, CHMILIST, CHEMSAFE, CIH, CSCHEM, CSHB, DETHERM*, DIFFR*, EWRASE, EKCOMPHATZ, GWOMPHIT, BNOOMPHIT, B

USPAT2, USPATFULL, VTB (*File contains numerically searchable property data)

Other Sources: DSL**, EINECS**, TSCA**

(**Enter CHEMLIST File for up-to-date regulatory information)

**PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT **

2492 REFERENCES IN FILE CA (1907 TO DATE) 20 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA

2493 REFERENCES IN FILE CAPLUS (1907 TO DATE)
23 REFERENCES IN FILE CAOLD (PRIOR TO 1967)

L99 ANSWER 8 OF 8 REGISTRY COPYRIGHT 2004 ACS on STN RN 76-32-4 REGISTRY

CN 3-Oxabicyclo[3.2.1]octane-2,4-dione, 1,8,8-trimethyl- (9CI) (CA INDEX NAME)

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WILLS 09/963800 5/19/04 Page 30
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OTHER CA INDEX NAMES: CN Camphoric anhydride (6CI, 7CI, 8CI) OTHER NAMES:

OTHER NAMES: CN (±)-Camphoric anhydride CN dl-Camphoric anhydride

CN DL-Camphoric anhydride CN NSC 4559 CN NSC 60293

CN NSC 657821 FS 3D CONCORD DR 595-30-2 MF C10 H14 03

USPATFULL ('File contains numerically searchable property data)

Other Sources: EIMEGS**
[**Enter CHEMAIST File for up-to-date regulatory information]

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**PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT **

117 REFERENCES IN FILE CA (1907 TO DATE)
5 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA

117 REFERENCES IN FILE CAPLUS (1907 TO DATE) 10 REFERENCES IN FILE CAOLD (PRIOR TO 1967)

0 REFERENCES IN FILE CAOLD (PRIOR TO 1967)